

## Chemistry 506: Allied Health Chemistry 2

### Chapter 20: Bioenergetics

#### Energy Generation in the Cell

Introduction to General, Organic & Biochemistry, 5<sup>th</sup> Edition by  
Bettelheim and March: Chapter 20, Pages 641-664

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1A Section(s) 20.1/2 Introduction & Mitochondria

## ➤ Metabolism

➤ All of the chemical reactions in a cell

## ➤ Catabolism

➤ The chemical reactions in the cell that break complex molecules down

## ➤ Anabolism

➤ The chemical reaction in the cell that build complex molecules

## ➤ Complexity

➤ Thousands of interrelated compounds, reactions, and enzymes

➤ All under detailed feedback and control

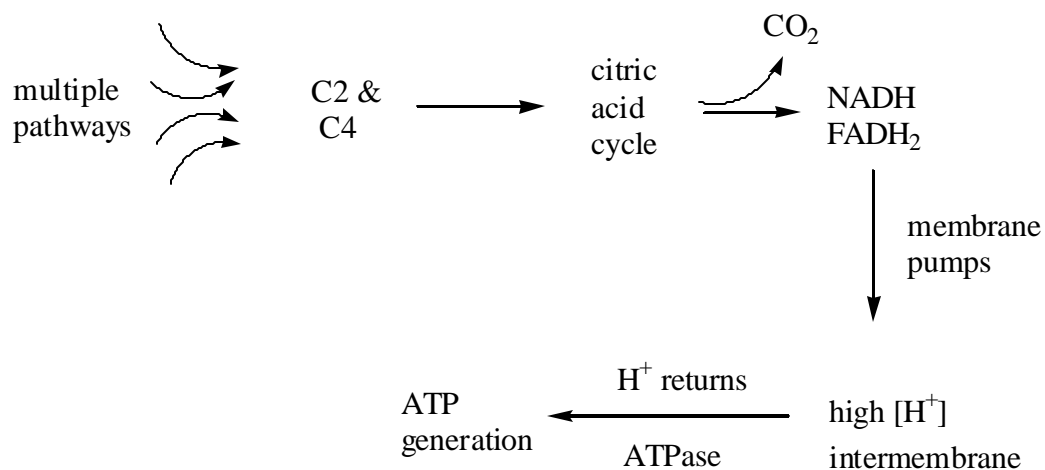
➤ Ultimately governed by the DNA and the cell's responses to the environment

➤ Energy Generation in the Cell

➤ General Process

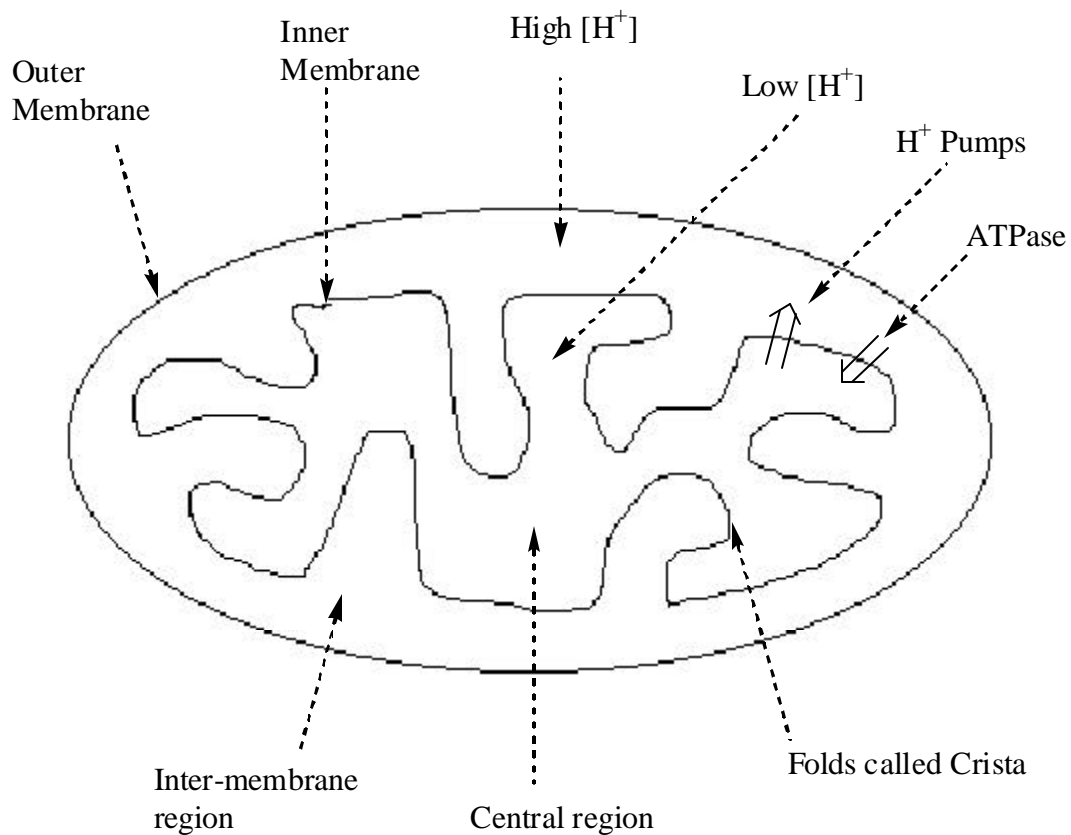
➤ Overall Catabolic Pathway

➤ Figure 20.1 on page 643



➤ Structure of **Mitochondrion**

➤ Figures 20.2 and 20.3 on pages 644 and 645



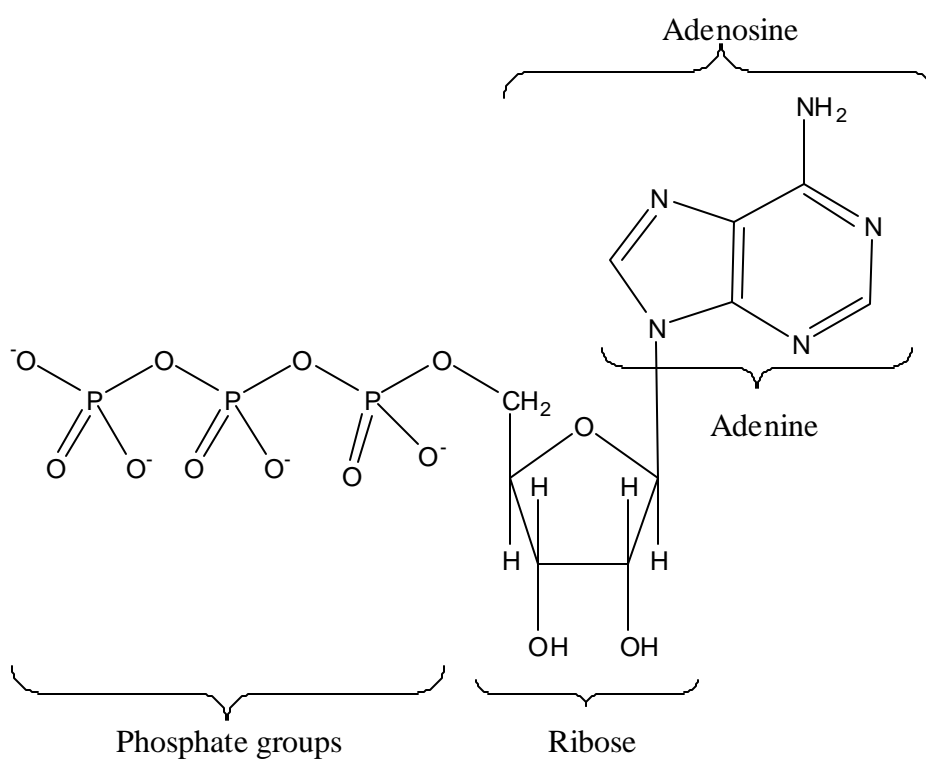
- Summary of Process
  - Multiple “food” molecules get converted into a small number of common C2 and C4 molecules
  - These C2/C4 molecules enter the center of the mitochondria where they are “processed” by the citric acid pathway
  - The citric acid pathway gives  $H^+$  and  $e^-$  which are used to generate NADH and  $FADH_2$ 
    - These are  $e^-$ ,  $H^+$ , and energy carrier molecules
  - These are used by proteins on the inner mitochondrial membrane to pump  $H^+$  ions from the center to the inter-membrane region
    - This gives a proton gradient
      - This proton gradient drives protein reactions on the inner membrane which allow them back into the center of the mitochondrion which simultaneously using their energy to generate ATP from ADP

1B Section(s) 20.3 Common Catabolic Molecules

- $P_i$ -AMP-ADP-ATP Path
  - Inorganic Phosphate,  $P_i$ ,  $H_2PO_4^-$  (charge depends on pH)
  - Adenosine Monophosphate, AMP
  - Adenosine Diphosphate, ADP
  - Adenosine Triphosphate, ATP
    - ATP is the highest energy

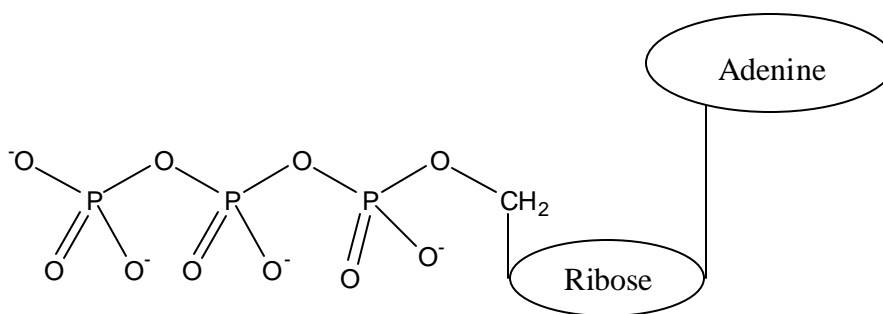
➤ Structure of ATP Molecule

- Page 646 of the text
- Adenine, Ribose, Adenosine, and Phosphate moieties
- ADP and AMP structures



➤ Phosphate Bonds

- ATP is “energy currency” of the cell
- “high energy bonds” vs. convertible energy



➤ Hydrolysis

➤ Hydrolysis of ATP



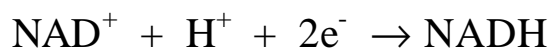
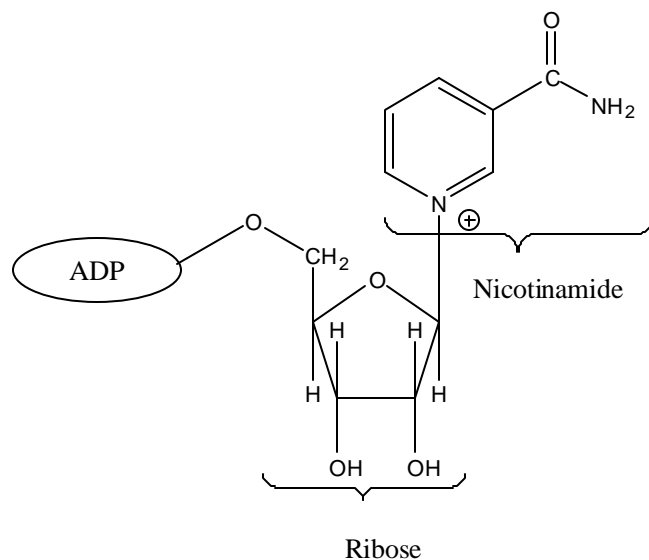
➤ Hydrolysis of ADP



- Normally ADP not hydrolyzed by cells



- Nicotinamide Adenine Dinucleotide, NAD
- Figure 20.6 on page 648
- Often Referred to as  $\text{NAD}^+$  to reflect the charge on the Nicotinamide base
- Is a Coenzyme
- Contains ADP, Ribose, and a Nicotinamide Base



- Thus NADH carries  $2\text{e}^-$ , a proton, and energy to where it is needed

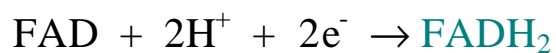
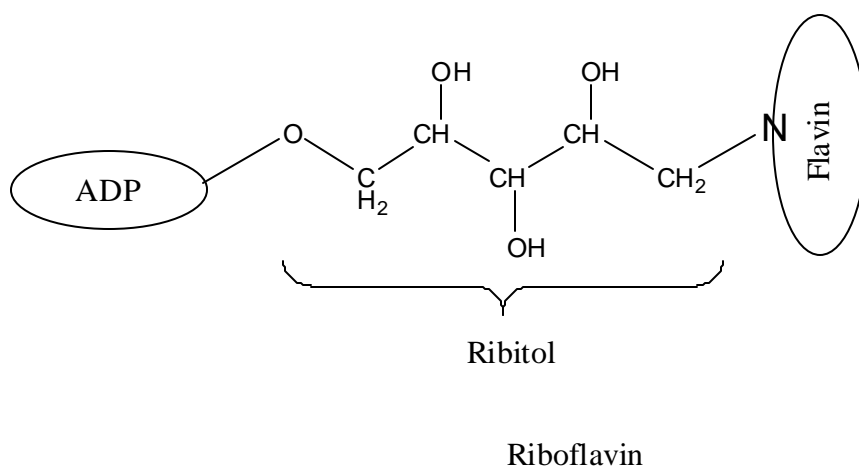
➤ **Flavin Adenine Dinucleotide, FAD**

➤ Figure 20.6 on page 648

➤ Is a **Coenzyme**

➤ Contains **ADP**, **Ribitol** (a straight chain **sugar**), and **Flavin**

➤ The latter two groups making up **Riboflavin** (the **vitamin**)



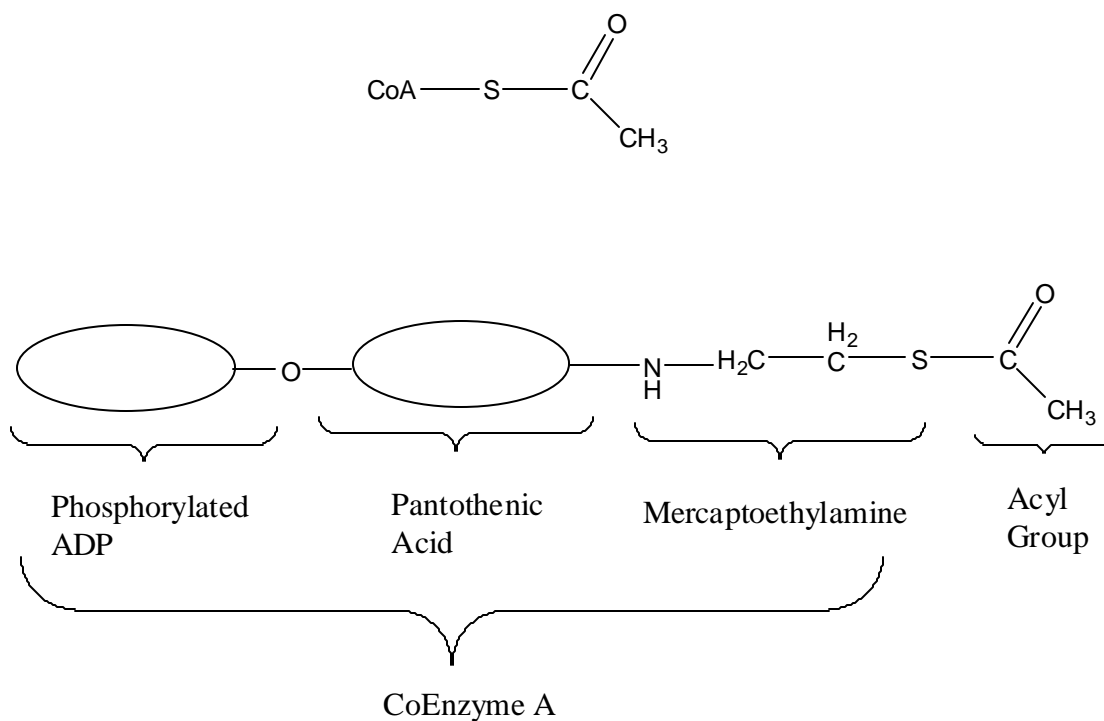
➤ Thus FADH<sub>2</sub> carries 2e<sup>-</sup>, two **protons**, and **energy** to where it is needed

➤ Acetyl CoA

➤ Figure 20.7 on page 649

➤ Transports C2 units (acyl groups)

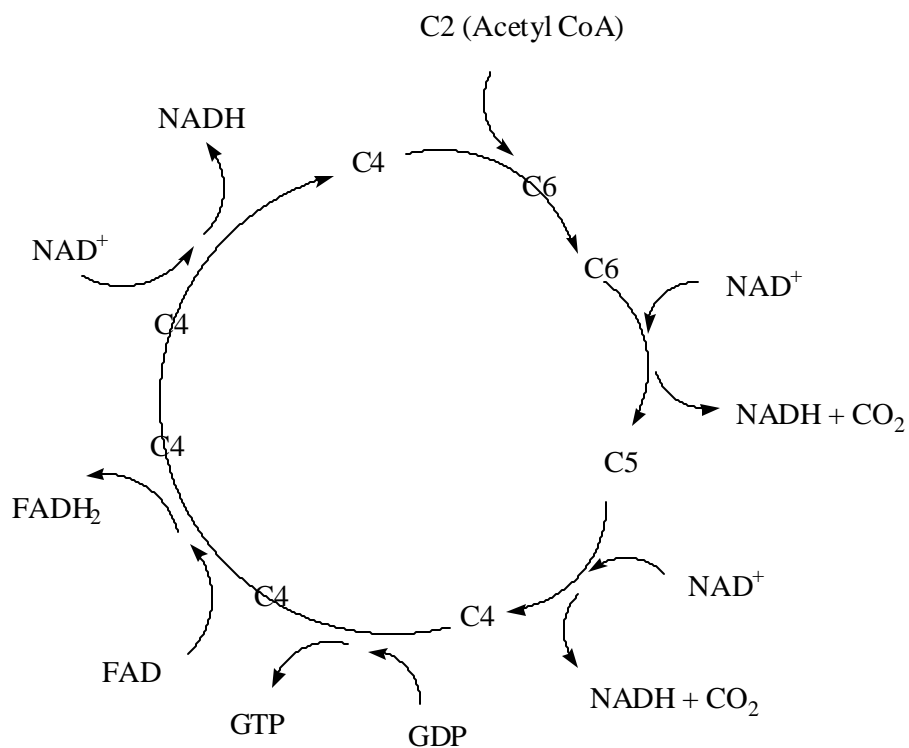
➤ Often written as  $\text{CH}_3\text{-CO-S-CoA}$  or Acyl-CoA



➤ Notice the overall similarity in the structures of ATP, NADH, FADH<sub>2</sub>, and Acetyl CoA

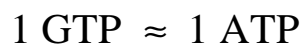
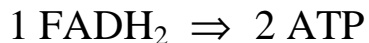
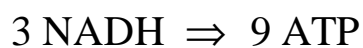
1C Section(s) 20.4 Citric Acid Cycle

- Also known as **Krebs Cycle** and **Tricarboxylic Acid Cycle**
- Figure 20.8 on page 650
- Overall **Molecular Flow**
  - 8 different chemicals
  - 8 different sets of **enzymes**
  - Takes place in the center of the **mitochondrion**
  - **C2 fragments** enter the cycle as **Acetyl CoA**



➤ Overall Energy Flow

- Produces two  $\text{CO}_2$ , three  $\text{NADH}$ , one  $\text{FADH}_2$ , and one  $\text{GTP}$  per cycle
- $\text{GTP}$  is Guanidine Triphosphate ( $\text{ATP}$  like)



1D Section(s) 20.5/6/7 ATP Synthesis

## ➤ Proton Pumps

- Flavo Protein, FeS Protein, Quinone Enzyme Complex
- Sited on the inner mitochondrial membrane
- Use NADH and FADH<sub>2</sub> to pump H<sup>+</sup> into the inter-membrane space
- This generates the proton gradient

## ➤ ATPase

- An enzyme on the inner mitochondrial membrane
- Allows H<sup>+</sup> to flow back into the central membrane cavity
- H<sup>+</sup> flow mechanically coupled to ATP generation



## ➤ Net Results

Each NADH ⇒ 3 ATP

Each FADH<sub>2</sub> ⇒ 2 ATP

1E Section(s) 20.8 Uses of Energy in Cells

- Molecular Synthesis
  - Anabolic Pathways
  
- To Generate Gradients via Active Pumps
  - $H^+$ ,  $K^+$ , etc.
  
- Mechanical Energy
  - Muscles
  - Molecular motors
  
- Heat

**Problems:** 20.1 to 20.49

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